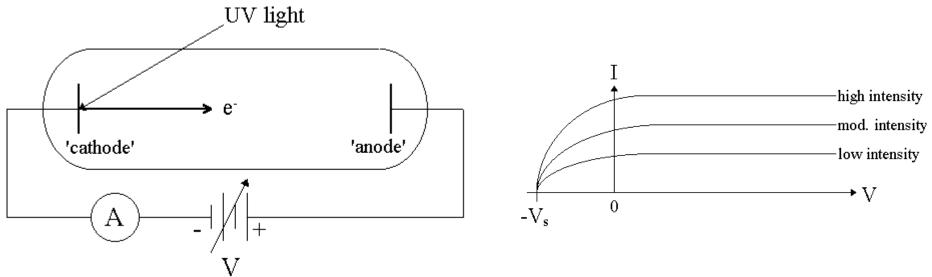
Wave properties for light: interference, diffraction, waveguide

But certain things cannot be explained by wave nature of light.

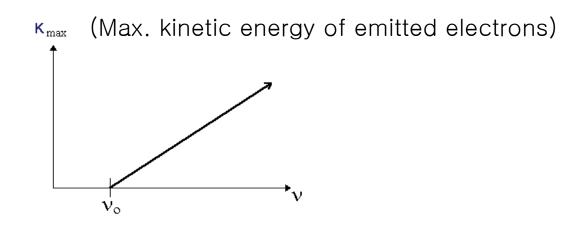
Photoelectron effects: electron emission when light shines on metal



- Amount of emitted electrons depends on light intensity

- Same minimum voltage for current flow regardless of light intensity
- → Same max. kinetic energy of emitted electrons regardless of light intensity





This results cannot be explained by wave nature of light.

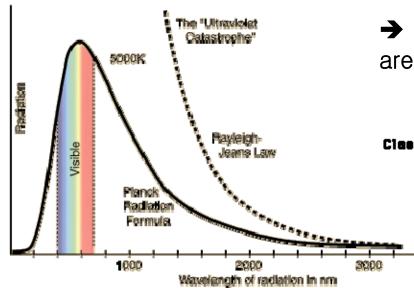
Einstein's explanation: Light delivers energy in chunks (photon)

$$E_{photon} = hv \cong \frac{1.24}{\lambda[\mu m]} eV$$
  
(h: Planck's constant =6.63x10<sup>-34</sup> Joule-sec)



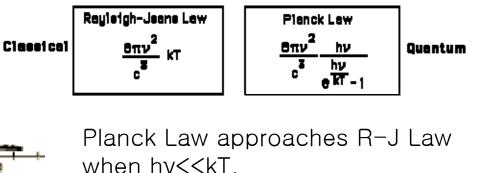
Consider the spectrum of light emission from an heated object (Thermal emission) (heat  $\rightarrow$  oscillatrion of charges inside the object => EM emission)

Very detailed analysis is possible for "black body" radiation (Rayleigh-Jeans Law).
(black body: object that absorbs 100% of incoming EM radiation => 100% emission)
→ Max. EM emission from a heated object at a given temperature)

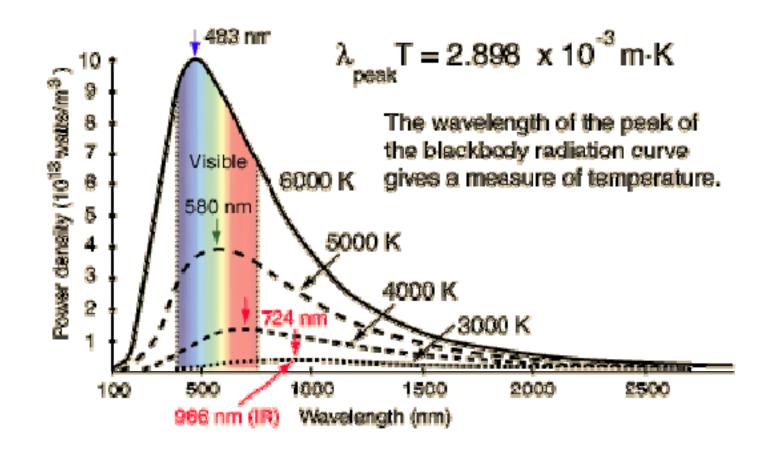


Rayleigh-Jeans law did not make sense at high frequencies.

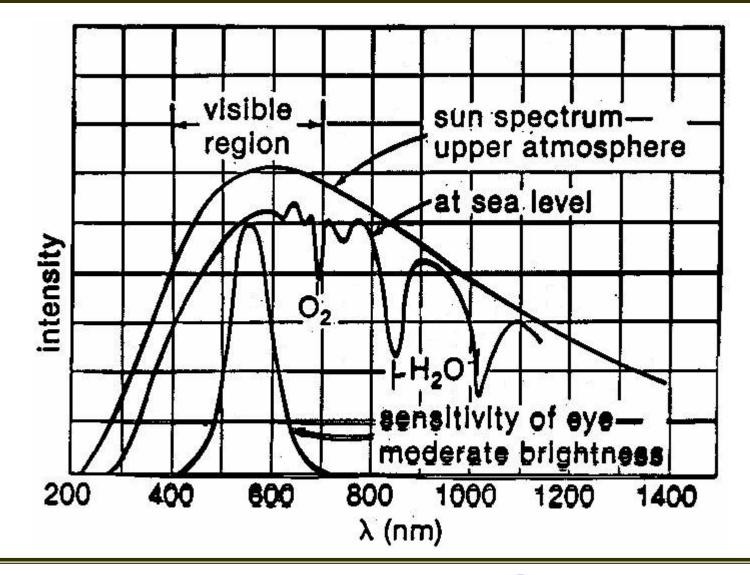
→ Planck suggested that EM energies are quantized (photon)  $E_{photon} = hv$ .

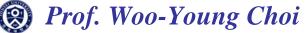




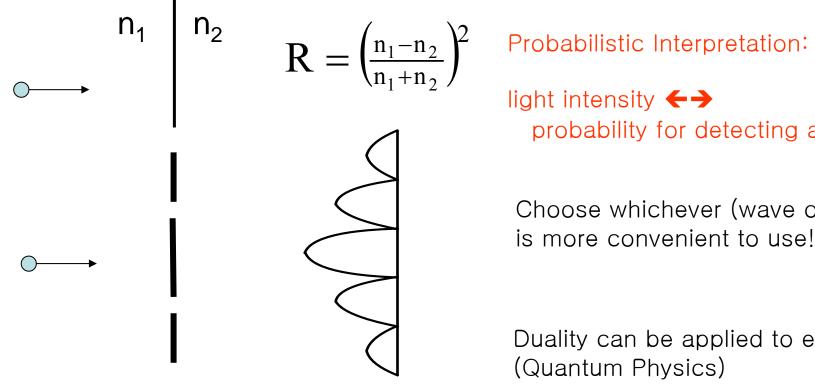








How can photons explain what we have learned: Reflection, interference, waveguide, ...



probability for detecting a photon

Choose whichever (wave or photon) is more convenient to use!

Duality can be applied to everything (Quantum Physics)

